Efficiency of Carbonyl Iron in Prevention of Anaemia in Piglets

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Abstract


The aim of this study was to evaluate efficiency of elemental iron preparation in the form of carbonyl iron in prevention of iron deficiency in piglets. The piglets in group I (n = 14) were given 210 mg of carbonyl iron orally at the age of 3 days. The piglets in group II (n = 15) received 210 mg of carbonyl iron orally on days 3 and 9. In group III (n = 14) the piglets were injected i.m. with 200 mg Fe3+ as iron dextran. Fourteen days after birth, haemoglobin concentration in group I started to decrease and the piglets developed anaemia. In group II, at the age of 28 days, Hb dropped below 80 g/l and the piglets developed anaemia. Under conditions of this trial, the oral administration of carbonyl iron did not prevent development of iron deficiency in piglets.

Anaemia, haemoglobin, packed cell volume, neonatal period, iron dextran, oral administration

The iron deficiency is a well known problem in swine production. The piglets are born with limited reserves of iron (ca 50 mg) (Venn et al. 1947). Without any preventive administration of iron the piglets develop anaemia within 10 - 14 days after birth (Egeli et al. 1998). The most common approach to prevention is i.m. administration of 200 mg Fe3+ in the form of iron dextran (Dilov and Chakurov 1984), although this method is not always without risks. The cases of acute toxicosis after iron dextran injection were described (Süveges and Glávits 1976; Kolb and Hoffmann 1989). An alternative to this method is oral administration of iron salts. Iron fumarate (Svoboda and Drábek 2002), iron lactate (Kotrbáček 2001; Svoboda et al. 2004) and iron sulphate (Schmitz and Müller 1971) have been used in swine production. Only iron salts with bivalent iron are used, because Fe2+ is up to 16 times better absorbed than Fe3+ (Dietzfelbinger 1987). Another possibility for oral iron administration represents elemental uncharged iron. The elemental iron preparation known as carbonyl iron powder has been developed (Sacks and Houchin 1978). “Carbonyl” does not refer to the composition of the iron particles but rather to the manufacturing process in which the controlled heating of vaporized iron pentacarbonyl leads to deposition of uncharged, elemental iron as submicroscopic crystals that form microscopic spheres of less than 5 µm in diameter. The preparation is more than 98% pure (Sacks and Houchin 1978; Gordeuk et al. 1986). Carbonyl iron has been shown to be absorbed and utilized for haemoglobin synthesis both in experimental animals and in humans (Sacks and Houchin 1978; Gordeuk et al. 1987). Carbonyl iron has been used for iron supplementation after blood transfusions (Gordeuk et al. 1990; Bianco et al. 2002). The carbonyl iron is less toxic compared to iron salts and has the advantage of decreased or absent risk of poisoning if accidentally ingested by children (Gordeuk 1987). To the authors’ knowledge, carbonyl iron has not yet been tested for the prevention of iron deficiency in suckling piglets.

Materials and Methods

Experimental design

A total of 53 piglets were used in the study. The piglets were divided into 4 groups using split litters, i.e. each litter was divided into four different groups. The carbonyl iron preparation (98% of elemental iron,
Superferrin, Favea s.r.o.) was used in our study. One capsule contained 14 mg of carbonyl iron and 200 µg of folic acid.

The piglets in group I (n = 14) were given 210 mg of carbonyl iron orally (15 capsules, 3 mg of folic acid) at the age of 3 days. At the age of 14 days they were treated with iron dextran (i.m., 200 mg Fe³⁺). The piglets in group II (n = 15) received 210 mg of carbonyl iron orally (15 capsules, 3 mg of folic acid) on day 3 and 9. In group III (n = 14) the piglets were injected i.m. with 200 mg Fe³⁺ in the form of iron dextran (Ferridextran 10%, Spofa a.s., Praha) and were given orally 3 mg of folic acid (Pharam Nord Praha, s.r.o.). The piglets in group IV (n = 10) did not receive any iron treatment until the age of 21 days. At this age they were treated with iron dextran (i.m., 200 mg Fe³⁺).

The creep feed (Maxi creep, Slavkovské krmné smûsi a.s., Marefy, iron content 225 mg/kg Fe) was offered to piglets from day 7 to day 35. The piglets were weaned at the age of 28 days.

Sampling
Blood was collected on days 3, 9, 14, 21, 28 and 35 from vena cava cranialis of the piglets. EDTA (ethylenediaminetetraacetic acid) was used as anticoagulant for the haematological examination. Heparin was used as anticoagulant for determination of iron concentration in blood plasma.

Haematological analysis
Haematological examination included: haemoglobin concentration (Hb), packed cell volume (PCV), red blood cell count (RBC), mean corpuscular volume (MCV), mean corpuscular haemoglobin (MCH) and mean corpuscular haemoglobin concentration (MCHC). These indices were determined by haematological analyzer Celtac Alfa (Nihon Kohden).

Iron concentration
Iron concentration in blood plasma (Fe) was determined photometrically, measuring iron complex with ferrozin according to Suwansaksri et al. (2003) (Iron liquid 917, Roche Diagnostic, Manheim, Germany).

Statistical analyses
The results were evaluated statistically by Kruskal Wallis ANOVA. The results are presented as mean values and standard deviation of each index in Figs 1 - 8. Values with *p < 0.05, **p < 0.01 express significant differences between iron dextran (group III) and the two carbonyl iron groups (group I, group II).

![Fig. 1. Haemoglobin concentration](image)

Results
The administration of 210 mg of carbonyl iron (group I) at the age of 3 days (day 3) resulted in a significant increase of Hb, Fe, PCV, MCV and MCH on day 9. Yet, on day 14 the above mentioned indices decreased and were significantly lower compared to iron dextran group \( (p < 0.01) \). The low values of Fe on day 14 indicate development of subclinical anaemia. The Hb values on days 21 and 28 were found to be lower than 80 g/l (anaemic limit). Therefore the piglets were given i.m. 200 mg Fe³⁺ as iron dextran on day 28. The body weight of piglets was significantly lower compared to iron dextran group.
**Fig. 2.** Iron concentration in blood plasma

**Fig. 3.** Packed cell volume

**Fig. 4.** Red blood cell count
After a double dose of 210 mg of carbonyl iron on days 3 and 9 (group II), the red blood cell indices and Fe were found to be comparable to those of the iron dextran group until the age of 14 days. From day 14 until the end of the trial Hb, Fe, PCV, MCV and MCH showed decreasing tendency and on day 28 Hb, PCV (both, $p < 0.01$), RBC ($p < 0.05$) and Fe ($p < 0.01$) were significantly lower than in the iron dextran group. On day 28 Hb values were lower than 80 g/l. No differences in MCHC were found among the groups during the trial.

The development of red blood cell indices and Fe in the group of piglets that did not receive any iron until the age of 21 days (group IV) was as expected, and the development of anaemia and decreased body weight of piglets were recorded.

Discussion

The anaemic limit, i.e. the point when iron deficiency gives clinical symptoms of anaemia, is set by most authors at a haemoglobin concentration below 80 g/l (Furugouri 1975; van Kempen 1987). From our results it can be concluded that neither a single nor a double dose of 210 mg of carbonyl iron did prevent development of iron deficiency in piglets. Compared to good efficiency of i.m. administration of 200 mg Fe$^{3+}$ as iron dextran, it is evident that utilization of carbonyl iron was much lower. This could be explained by specific mechanism of carbonyl iron absorption.

The combination of carbonyl iron and folic acid was used in the study. The folic acid has an important role in erythropoiesis. Erythroblasts require folic acid for proliferation during their differentiation (Koury and Ponka 2004). The same amount of folic acid was also given to piglets that were injected intramuscularly with iron dextran.

The conversion of particulate carbonyl iron to soluble ionized iron by stomach acid is a prerequisite for absorption and is limited by the rate of gastric acid production (Huebers et al. 1986). With ferrous iron salts, all the iron is in a soluble ionized form and is potentially available for absorption. With carbonyl iron, however, only that fraction solubilized by gastric acid is available for absorption. According to Swain et al. (2003) the bioavailability of carbonyl iron in rats is 64% of that of ferrous sulphate. Devasthali et al. (1991) found in humans that the bioavailability of carbonyl iron was about 70% of that of ferrous sulphate. These data show higher bioavailability of iron salts. This is also in agreement with our previous study where we have demonstrated the same efficiency of a double dose of 200 mg Fe$^{2+}$ in the form of iron fumarate as that of i.m. administration of 200 mg Fe$^{3+}$ in the form of iron dextran (Svoboda and Drábek 2002).
In addition, the rate of solubilization is restricted by the rate of gastric acid production (Huebers et al. 1986). Swain et al. (2003) showed that solubility of carbonyl iron depends on gastric pH, i.e. it is less soluble at higher pH. It should be taken into account that the previously cited bioavailabilities of carbonyl iron (Devasthali et al. 1991; Swain et al. 2003) were obtained at gastric pH below 2. Because of relatively high pH of gastric content in suckling piglets, the negative effect of gastric pH on carbonyl iron utilization could be expected.

It has been documented that gastric mucosa of newborn piglets has a secretory activity of hydrochloric acid production (Cranwell et al. 1976; Black et al. 1981; Sangild et al. 1995). However, this secretory activity has limited capacity in suckling piglets (Cranwell et al. 1976). The major contributor to acidification of the stomach content of suckling piglets is lactic acid produced by the fermentation process (Cranwell et al. 1976). Snoeck et al. (2003) documented values of gastric pH in suckling piglets ranging from 3.0 - 4.1. Similar data were obtained by Cranwell et al. (1976) who reported mean gastric pH 4.1 ± 0.9 at the age of 2 days and 4.1 ± 0.5 at the age of 10 days.

The relatively high gastric pH in suckling piglets can be explained by several factors. Firstly, the stomach of suckling piglets has a less developed capacity to secrete hydrochloric acid (Cranwell et al. 1976). Secondly, the sow’s milk is not strongly stimulating for this secretion, and has a considerable buffering capacity (Cranwell et al. 1976; Snoeck et al. 2004). Thirdly, lactic acid production from lactobacilli in the stomach of suckling piglets can partly or completely inhibit acidification by HCl (Cranwell et al. 1976).

It can be concluded that oral administration of carbonyl iron resulted in a significant haematological response but it did not prevent iron deficiency of piglets under the conditions of our trial. We suggest that more frequent administration would be necessary in order to achieve efficiency comparable to iron dextran; however, this is hardly feasible under practical conditions of pig production.

**Prevence nedostatku železa u sajících selat pomocí karbonylového železa**

Cílem studie bylo zjistit, jaká je účinnost karbonylového železa pro prevenci nedostatku železa u selat. Šelátům ve skupině I (n = 14) bylo aplikováno perorálně 210 mg karbonylového železa ve věku 3 dnů. Šelátům ve skupině II (n = 15) bylo podáno perorálně 210 mg karbonylového železa ve věku 3 a 9 dnů. Šelátům ve skupině III (n = 14) bylo aplikováno 200 mg Fe^{3+} ve formě dextranu železa. Čtrnáct dní po narození začala koncentrace hemoglobinu ve skupině I klesat a u selat se vyvinula anémie. Ve skupině II klesla koncentrace hemoglobinu ve věku 28 dnů pod 80 g/l a u selat se vyvinula anémie. Za podmínké této studie nezabránila perorální aplikace karbonylového železa vzniku anémie u selat.

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