

## Analysis of Goat Milk by Near-Infrared Spectroscopy

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### Abstract

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The objective of this study was to determine protein, fat, lactose, total solids, non-fatty solids contents, freezing point, titratable acidity and pH using Fourier transform near infrared spectroscopy (FT-NIR). Sixty samples of goat milk were used to calibrate the instrument by the partial least squares (PLS) method. The spectra were measured on the integration sphere in the reflectance mode with the use of a 0.1 mm wide transfectance cell. The following statistical values were obtained: correlation coefficient ( $R$ ) = 0.920 and standard error of calibration (SEC) = 0.094 for protein,  $R$  = 0.951 and SEC = 0.124 for fat,  $R$  = 0.997 and SEC = 0.011 for lactose,  $R$  = 0.940 and SEC = 0.260 for total solids,  $R$  = 0.873 and SEC = 0.159 for non-fatty solids,  $R$  = 0.935 and SEC = 0.003 for freezing point,  $R$  = 0.952 and SEC = 0.295 for titratable acidity and  $R$  = 0.835 and SEC = 0.057 for pH. The calibration models developed were verified by cross validation. The study showed that FT-NIR is a potentially useful technique for evaluating the composition of goat milk.

*Total solids, protein, fat and sugar content, freezing point, titratable acidity, pH, milk, composition*

The quality and composition of goat milk, which is a healthy, wholesome and easily digestible food, depends on a series of factors. These factors are interrelated especially with the breed characteristics, lactation phase, heredity, the animal's individuality, breeding practices, feeding practices, milking method and the health condition of the animals (Aganga et al. 2002).

Traditional methods for the quality assessment of food stuffs (flavour, composition, contamination) are too time-consuming and costly. Several authors thus focused on monitoring the basic food components with near infrared spectroscopy (NIRS). This method has been in use for about 40 years in food stuff analysis (Lee 2004).

NIRS has many properties that rank it among excellent food stuff analysis techniques. It is a physically non-destructive method that has several advantages as compared to the traditional methods. It is rapid, accurate, easily carried out, objective and it can give both quantitative and qualitative results in many component analyses (Ru and Glatz 2000; Jankovská and Šustová 2003). NIR spectroscopy is used mostly for the determination of the basic components, i.e. total solids, protein, fat and sugar content. Its applications, however, vary much more and include even the determination of sensory and physico-chemical variables, such as density, freezing point, the pH value etc. (Ru and Glatz 2000; Čurda et al. 2002).

The use of NIR spectroscopy for monitoring cow and goat milk indicators was focused on by many authors (Jankovská and Šustová 2003; Albanell et al. 1999). The first calibration for the determination of the basic goat milk components was done by Díaz-Carrillo et al. (1993) and Ru and Glatz (2000). Apart from determining the fat, protein,

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lactose and total solid contents and somatic cell counts in goat milk, NIR spectroscopy was used for establishing the total casein,  $\alpha$ -,  $\beta$ - and  $\kappa$ -casein contents (Díaz-Carrillo et al. 1993).

The objective of our study was to develop calibration models for determining the basic goat milk components using FT-NIR.

### Materials and Methods

Milk samples were collected from White Shorthaired breed goats, kept at a Czech farm. In order to develop calibration models, 60 bulk tank samples of raw goat milk were used. Sampling was done after the kids had been weaned, in the period from the end of April till the beginning of September 2006 at regular time intervals. At the farm, there were 75 goats in their 1<sup>st</sup> to 8<sup>th</sup> lactations, the average milk yield being 2–3 l of milk, the average yearly yield 600–800 l of milk. In the period between mid-May and mid-November, the goats could go on pasture. The feed ration was complemented with 0.5 kg of hay, a maximum of 1 kg of grain cereals, vitamin and mineral mixture and a salt block for licking. In the winter, the feed ration contained 3 kg of grass haylage, 1 kg of sugar beet silage, 1 kg of hay and a maximum of 1 kg of grain cereals, vitamin and mineral mixture and a salt block for licking. Milking was carried out on milking machines twice a day.

In the samples, the following indicators were determined: the contents of protein, fat, lactose, non-fatty solids - all of them on the Bentley 2500 machine (Bentley Instruments, Chaska, USA) based on the Czech National Standard No. 570536 (1999), total solids content (CSN ISO 6731, 1997), titratable acidity and pH levels (CSN 570530, 1972). The freezing point was determined based on the Czech National Standard No. 570538 (1998) using a Milk Cryoscope 4D2 machine (Advanced Instruments, Inc., Norwood, USA).

The milk samples were measured with a Nicolet Antaris Near-IR Analyzer (Thermo Electron Corporation, Madison, USA) in the spectral range of 10 000–4 000  $\text{cm}^{-1}$  with 100 scans. One spectrum analysis time ranged around 1.5 min. The spectra were measured on the integration sphere in the reflectance mode with the use of a transmittance cell 0.1 mm wide. The samples were warmed up to 40 °C, stirred up and after cooling down to 20 °C, they were transferred onto Petri dishes. The collected data were processed with the TQ Analyst version 6.2.1.509 software using the PLS (partial least square). The same samples were used for cross validation. The results were processed with statistical and graphic software STAT Plus (Matoušková et al. 1992).

### Results and Discussion

A total of 60 bulk tank samples of raw goat milk were used for the development of the calibration models. The ranges of the reference values for the monitored indicators (protein, fat, lactose, total solids, non-fatty solids contents, freezing point, titratable acidity and pH levels) are presented as standard errors of the mean (Table 1). Samples that displayed an error in the measured spectrum or for which the reference value was established inaccurately were removed using the Spectrum Outlier and Leverage (Table 2) diagnostic device.

The calibration models for all the monitored indicators were developed using a PLS algorithm. In the analyzed samples PLS used spectral as well as concentration information to determine latent variables (PLS factors) in a data set (Tsenkova et al. 2000). In all of the developed calibration models, the maximum number of PLS factors was set at 10. The highest factor count was found in lactose (10). The lowest PLS factor count (4) was

Table 1. Reference values

Variable	n	$x_{\min}$	$x_{\max}$	$\bar{x}$	SD
Protein [%]	60	2.33	3.41	2.81	0.23
Fat [%]	60	2.27	5.61	3.13	0.54
Lactose [%]	60	4.39	4.94	4.59	0.10
Total solids [%]	47	10.30	13.76	11.14	0.71
Non-fatty solids contents [%]	57	7.19	8.81	7.97	0.34
Freezing point [°C]	49	-0.599	-0.527	-0.554	0.010
Titratable acidity [SH]	49	4.60	8.20	5.61	0.84
pH	42	5.69	6.92	6.63	0.18

n – samples number,  $\bar{x}$  - arithmetic average,  $x_{\min}$  – minimum value,  $x_{\max}$  maximum value, SD - standard deviation

found in the protein content and freezing point. Jankovská and Šustová (2003) obtained 11 PLS factors for lactose content and 14 PLS factors for protein content. Also Díaz-Carrillo et al. (1993) used the PLS method for calibration of basic goat milk components but the samples were measured in transmittance mode. All of the calibration models were developed without spectrum derivation.

The individual calibration

Table 2. Calibration and validation results

Variable	n	PLS factors	calibration			validation		
			<i>R</i>	SEC	CCV [%]	<i>R</i>	SECV	PCV [%]
Protein [%]	55	4	0.920	0.094	3.37	0.888	0.111	3.99
Fat [%]	52	6	0.951	0.124	4.04	0.924	0.154	5.02
Lactose [%]	58	10	0.997	0.011	0.24	0.935	0.050	1.09
Total solids [%]	41	6	0.940	0.260	2.32	0.899	0.334	2.98
Non-fatty solids contents [%]	51	6	0.873	0.159	2.00	0.812	0.191	2.40
Freezing point [°C]	45	4	0.935	0.003	0.52	0.833	0.005	0.82
Titrate acidity [SH]	49	5	0.952	0.295	4.96	0.878	0.469	7.89
pH	41	7	0.835	0.057	0.85	0.703	0.076	1.14

n – samples number after removed outlying, *R* – correlation coefficient, SEC – standard error of calibration, SECV – standard error of validation, CCV – calibration coefficient of variation, PCV – prediction coefficient of variation

correlation coefficients (*R*) were obtained in the interval from 0.997 for lactose to 0.835 for pH with the standard errors of calibration (SEC) 0.011% and 0.057. The highest standard errors of calibration were found for titrate acidity at 0.295 SH (*R* = 0.952) and total solids at 0.260% (*R* = 0.940) (Table 2, Figs 1 - 8). Similar results were reported by Díaz-Carrillo et al. (1993) who monitored the contents of protein, total casein,  $\alpha$ -,  $\beta$ - and  $\kappa$ -casein, fat and lactose in goat milk using an NIR spectrometer. They found low standard errors and high correlation coefficients: the calibration correlation coefficient was 0.96 for protein, 0.96 for fat and 0.99 for lactose. Their study confirmed also the applicability of NIR spectrometry for determination of casein fractions, even though they were present at low concentrations. Šustová and Kuchtík (2007) showed that when milk samples are measured on Petri dishes, the NIR spectroscopy method is a useful technique for the prediction of fat, true protein, casein and lactose in goat milk. They obtained very high correlation coefficients of calibrations (fat 0.907, true protein 0.989, casein 0.890 and lactose 0.981).

For the development of a validation model using the cross-validation method, the same set of samples was used as for calibration. Cross-validation confirmed the reliability of the calibration model. The correlation coefficients of the validation (*R*) were found in the interval of 0.935 for lactose and 0.703 for pH with the standard errors of calibration and

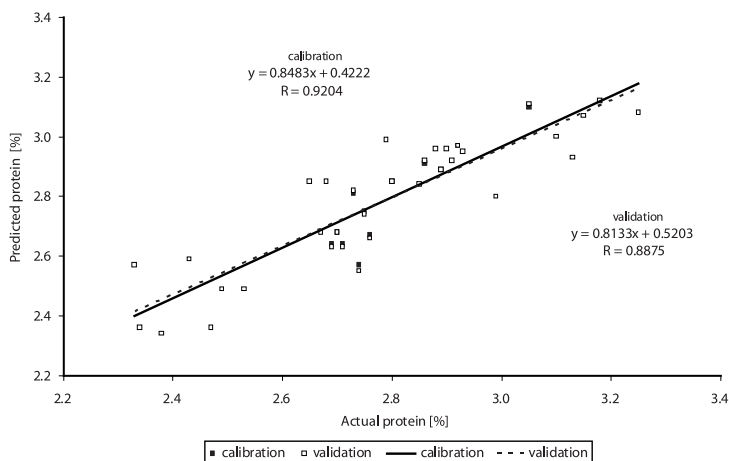


Fig. 1. Calibration and validation models of protein

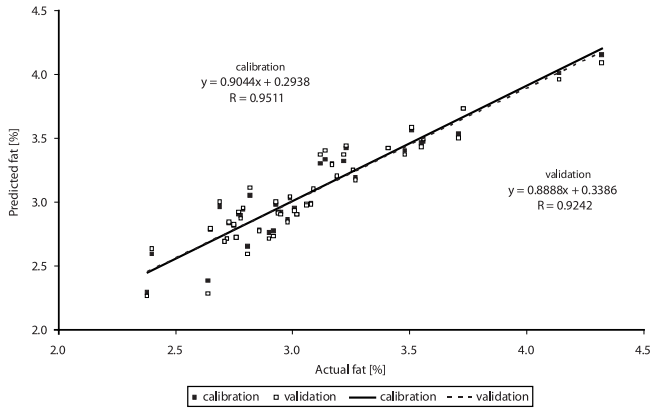


Fig. 2. Calibration and validation models of fat

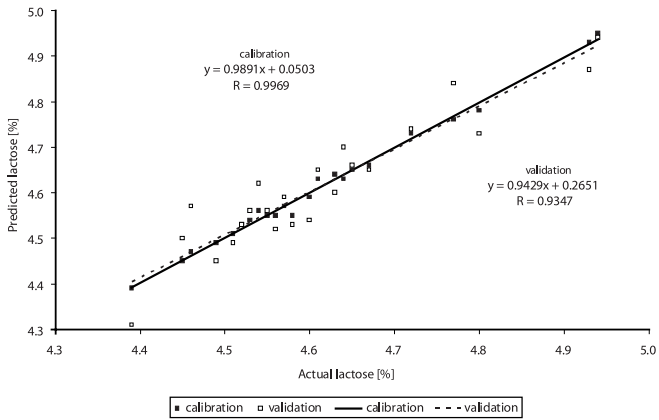


Fig. 3. Calibration and validation models of lactose

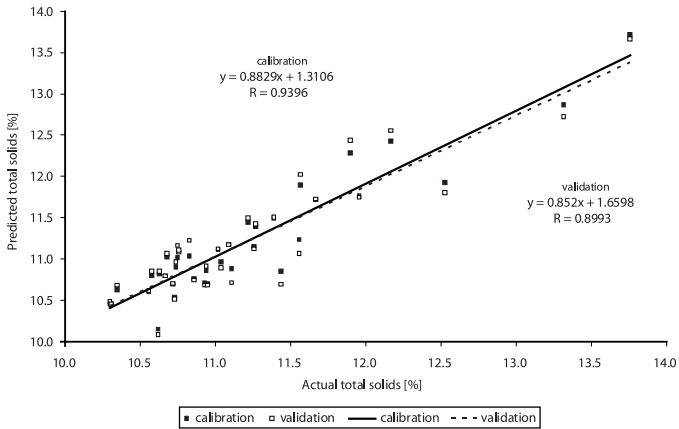


Fig. 4. Calibration and validation models of total solids

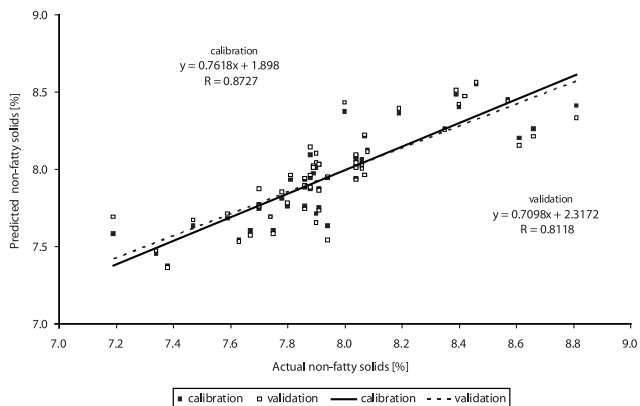


Fig. 5. Calibration and validation models of non-fatty solids

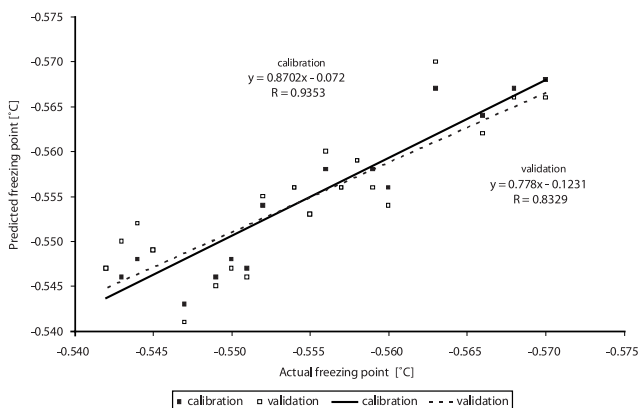


Fig. 6. Calibration and validation models of freezing point

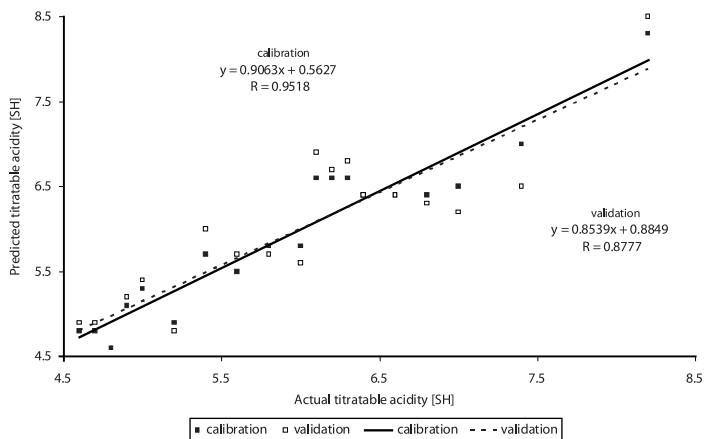


Fig. 7. Calibration and validation models of titratable acidity

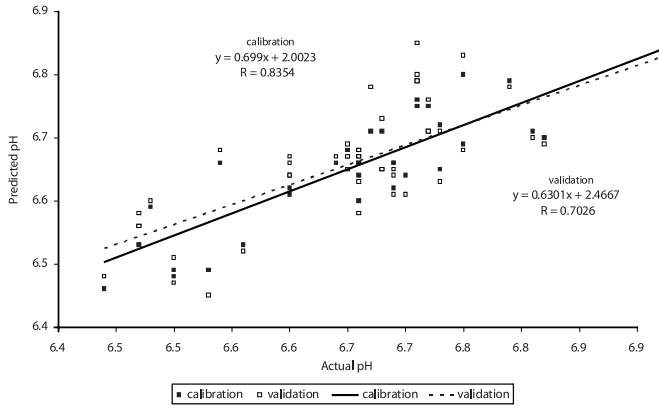


Fig. 8. Calibration and validation models of pH

validation (SECV) at 0.050% for lactose and 0.076 for pH. The highest SECV were found for titratable acidity at 0.469 SH ( $R = 0.878$ ) and for total solids at 0.334% ( $R = 0.899$ ) (Table 2, Figs 1 - 8).

The indirect method (the mid-infrared spectroscopy - analyzer Bentley) as a reference method for determination of protein, fat, lactose and non-fatty solids was used. This method can influence subsequent results of calibration models. Šustová et al. (2007) used the PROMILK instrument for the calibration of the FT-NIR spectrometer as a reference method for determination of true protein and casein in cow, ewe and goat milk. This method is based on the formation of an insoluble protein-dye complex, forming after the addition of excess Amido Black 10B to a buffered solution.

The error of the indirect method calibration for milk composition may be the reason of less robust models for protein and non-fatty solids contents. Less robust models were obtained for freezing point and titratable acidity.

Šustová et al. (2007) used FT NIR spectroscopy for determination of true protein and casein in goat milk. They obtained correlation coefficients of calibrations ( $R$ ) and standard errors of calibration (SEC) as follows: for true proteins  $R = 0.989$ , SEC = 0.06 and for casein  $R = 0.890$ , SEC = 0.12. The values of validations were for true protein  $R = 0.972$ , SECV = 0.09 and for casein  $R = 0.814$ , SECV = 0.16. These authors confirmed that near infrared spectroscopy enables an easy and quick control of milk composition, timely interventions in the production, and thus may improve the economics of dairy plants, in particular of cheese producers.

NIR reflectance technique for determination of the contents of protein, casein, total solids and somatic cell counts in goat milk was used also by Albanell et al. (2003). These authors performed analyses in unhomogenized and homogenized goat milk samples and they achieved high correlation coefficients which are comparable with our results. For homogenized milk, Albanell et al. (2003) found the following calibration and validation correlation coefficients: 0.98 and 0.97 for fat, 0.96 and 0.95 for protein and 0.94 and 0.93 for total solids. For unhomogenized milk, the values were: 0.98 and 0.97 for fat, 0.95 and 0.95 for protein and 0.95 and 0.95 for total solids. Šustová and Kuchtík (2007) found that results of calibration for somatic cells are not accurate (correlation coefficient of calibration 0.885 and correlation coefficients of validation 0.566).

Based on the assessment of the variable values of calibration coefficients of variation (CCV) and prediction coefficients of variation (PCV) (Jankovská and Šustová 2003; Albanell et al. 1999), very reliable models were developed (Table 2). The results were

processed statistically using the STAT Plus software (Matoušková et al. 1992). No significant differences ( $p = 0.05$ ) were found between the reference values and those calculated using FT-NIR.

Calibration models were obtained for determination of the contents of protein, fat, lactose, total solids, non-fatty solids, freezing point, titratable acidity and pH in the goat milk. The results were assessed based on the correlation between the reference and calibration equation calculated values and based on the standard errors of calibration and validation (SEC, SECV). For all of the monitored indicators, very reliable calibration models were developed. Cross-validation pointed out the applicability of NIR spectrometer for the determination of basic components. FT-NIR spectroscopy presents a viable technique for rapid goat milk analysis. The models for freezing point and titratable acidity were less robust therefore FT-NIR spectroscopy is suitable only for screening determination of these indicators.

### Analyza kozího mléka pomocí NIR spektroskopie

Cílem práce bylo využití blízké infračervené spektroskopie s Fourierovou transformací (FT-NIR) v kombinaci s metodou částečných nejmenších čtverců (PLS) pro stanovení obsahu bílkovin, tuku, laktózy, sušiny, tukuprosté sušiny, bodu mrznutí, titrační kyselosti a pH u 60 vzorků kozího mléka. Spektra byla naměřena v modu reflektance s transflektanční kyvetou o tloušťce vrstvy 0,1 mm ve spektrálním rozsahu 10000 - 4000  $\text{cm}^{-1}$  se 100 scany. Pro sledované ukazatele byly vytvořeny kalibrační modely, které byly statisticky zhodnoceny na základě korelačních koeficientů ( $R$ ) a směrodatných odchylek kalibrace (SEC). Pro bílkovinu byly zjištěny hodnoty  $R = 0,920$  a  $\text{SEC} = 0,094$ , pro tuk  $R = 0,951$  a  $\text{SEC} = 0,124$ , pro laktózu  $R = 0,997$  a  $\text{SEC} = 0,011$ , pro sušinu  $R = 0,940$  a  $\text{SEC} = 0,260$ , pro tukuprostou sušinu  $R = 0,873$  a  $\text{SEC} = 0,159$ , pro bod mrznutí  $R = 0,935$  a  $\text{SEC} = 0,003$ , pro titrační kyselost  $R = 0,952$  a  $\text{SEC} = 0,295$  a pro pH  $R = 0,835$  a  $\text{SEC} = 0,057$ . Vytvořené kalibrační modely byly následně ověřeny pomocí cross-validace. Ze studie vyplývá, že FT-NIR spektroskopie představuje vhodnou techniku pro rychlou analýzu základních složek kozího mléka.

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